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© OCR 2015 A Level in Chemistry A General Information . Molar gas volume = 24.0 dm³ mol⁻¹ at room temperature and pressure, RTP Avogadro constant, N_A = 6.02 × 10²³ mol⁻¹ Specific heat capacity of water, c = 4.18 J g⁻¹ K⁻¹ Ionic product of water, K_w = 1.00 × 10⁻¹⁴ mol² dm⁻⁶ at 298 K 1 tonne = 106 g Arrhenius equation: k = Ae^{-Ea/RT} or ln k = -E

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A-level Chemistry Specimen mark scheme Paper 1

A-level Chemistry (7405/1) Paper 1: Inorganic and Physical Chemistry . Specimen 2015 v0.5 Session 2 hours . Materials . For this paper you must have: • the Data Booklet, provided as an insert • a ruler • a calculator. Instructions • Answer . all . questions. • Show . all . your working. Information • The maximum mark for this paper ...

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Cambridge International Advanced Subsidiary and Advanced Level in Chemistry (9701) For use from 2016 in all papers for the above syllabus, except practical examinations. CSTxxx ** 2 Cambridge Pre-U Revised Syllabus Contents: Tables of Chemical Data Page no. 1 Important values, constants and standards 3 2 Ionisation energies (1st, 2nd, 3rd and 4 ...

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Data sheet H432 H032 - PDF 2MB; Periodic table, elements and physical chemistry H432/01 - Sample question paper and mark scheme. PDF 2MB; Synthesis and analytical techniques H432/02 - Sample question paper and mark scheme. PDF 3MB; Unified chemistry H432/03 - Sample question paper and mark scheme. PDF 2MB

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AS/A Level Chemistry

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