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The salts of strong acids and weak bases give acidic solution having pH less than 7. Example, NH_4Cl , Ammonium Chloride will have pH less than 7. The salts of weak acids and strong bases give basic solution having pH more than 7. Example, Na_2CO_3 , Sodium Carbonate will have pH more than 7. Question 7.

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Acids And Bases Answer Key - Teacher Worksheets

Answer. (a) HCl is acid and NaOH is base whose combination forms the common salt. Its formula is NaCl (Sodium chloride). It is obtained from sea water. (b) Rock salt is the common name for the mineral “ halite ” . Its chemical formula is NaCl. It may be white or light blue or yellow depending upon impurities present in

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it.

Acids Bases and Salts Chapter Wise Important Questions ...

Acids, Bases and Salts SCIENCE AND TECHNOLOGY Notes MODULE - 2

Matter in our Surroundings Acid Base

INTEXT QUESTION 8.1 1. Put the

following substances in acid or base

bottle. (a) Milk of magnesia (b) gastric

juice in humans (c) soft drinks (d) lime

water (e) vinegar (f) soap 2. What will

happen if you add a drop of the

following on a cut ...

Notes ACIDS, BASES AND SALTS

GCSE Chemistry Acids, bases and salts

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NCERT Solutions for Class 10 Chapter 2 Acids, bases and salts are prepared to help students in their exam preparation. This solution provides you with answers to the questions provided in the NCERT Class 10 textbooks.

NCERT Solutions for class 10 chapter 2 Acids, bases and salt

Many of the acids that we do not consume in the household are used in the laboratories and industries, which include an acid such as HCl , H_2SO_4 etc. and bases such as NaOH , KOH etc. When these acids and bases are mixed in the right proportions, the neutralization reaction thus results in the formation of salt and water. Some naturally occurring salts found in nature include NaCl and KCl etc in seawater and natural rock deposits. In

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this section, we will read more about acid, base and salt ...

Acids, Bases, and Salts - Introduction, Dissociation ...

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Chapter 19 Acids Bases Salts Answer Key

answer choices The substance has a pH value of "2". The substance is a base. The substance has a pH value of "7".

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Acids/Bases and Salts | Chemistry Quiz - Quizizz

Acid + Base Salt + Water (heat is evolved) Question 10. Name three acids used in the laboratory. Answer: Acids which are mostly used in laboratory as below. Hydrochloric acid (HCl) Sulphuric acid (H_2SO_4) Nitric acid (HNO_3) Question 11. Paheli is suffering from indigestion due to acidity.

Acids, Bases and Salts Class 7 Extra Questions Science ...

MBE = effective concentration of base

NOTE: MB = concentration of base

nOH = number of hydroxides

Chemistry: Form WS9.5.1A Name

_____ ACIDS, BASES, AND SALTS Date

_____ Period _____ Titration

Titration is a process that uses a

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neutralization reaction to determine the concentration of an acid or a base.

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Evan P Silberstein Answer Key 2002
When acids react with bases they usually form salts. How conversant are you with the solutions of this reaction. Take up the simple quiz below and see if indeed you

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understood the topic on Acids, Bases and Salt. All the best as you tackle it and remember to always revise.

Acid, Bases And Salts - ProProfs Quiz
Similarity between Acids and Bases
Both acids and base react with water. They produce ions in water Both acids and bases acts as electrolytes, so are good conductors of electricity.

Revision Notes on Acids, Bases and Salts - Askiitians

11. Mixing an acid or base with water results in decrease in the concentration of per unit volume. This process is called 12. Among HCl, H₂SO₄ and CH₃COOH, is a weak acid. Answers. 1. blue, red 2. more/greater 3. CaSO₄, Cl₂, H₂O 4. 7 or seven 5. Water of crystallisation 6. Sodium chloride 7. acidic, less 8.

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baking soda 9. acid (HCl) 10. H + 11.

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