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9 Group 2, the Alkaline Earth Metals Answers to practice questions eases indigestion 4 (b) M1 Decrease in T decreases the energy of the particles / ions / H⁺ / molecules M2 (also scores M1) Decrease in the number of / less particles / ions / H⁺ / molecules with E \rightarrow E Act or E \rightarrow minimum energy to react M3 Few(er) / Less effective / productive /

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Group 2: The Alkaline Earth Metals Lab. Pre-Lab Discussion. The elements in Group 2 of the periodic table are called the alkaline earth elements. Like the elements in Group 1 (the alkali metals), the elements in Group 2 are chemically active and are never found in nature in the elemental state. Like all members of a group, or family, the elements in Group 2 share certain common characteristics.

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Group 2: The Alkaline Earth Metals Lab

Conclusion: I came to the conclusion that Alkaline Earths should appear as (Mg, Ca, Sr, Ba) from top to bottom respectively in the periodic table. I also concluded that Halogens should appear (Cl, Br, I) from top to bottom respectively in the periodic table. I also found that Unknown #8 = BaCl₂. Lab Notebook:

The Alkaline Earths and the Halogens: Two Families in the ...
As something melts, its adjustments from a sturdy to a liquid—and not making use of a replacement in temperature. So density is incomprehensible. The density is beside the point—that in the time of basic terms impacts the quantity of substance with a given mass. fairly the incredibly one mandatory is b) the

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mass, yet you in addition to might want the Latent heat temperature* of Fusion for the ...

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Background: Elements within a given column of the periodic table tend to have similar properties due to their similar valence electron configurations. Because of this, columns of elements are often...

Alkaline Earths & Halogens Experiment (1 of 2)

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Alkaline Earth Metals Lab Answer Key

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In this lab, you will compare the abilities of the alkaline earth metals to form precipitates as a result of a chemical reaction.

Materials. White paper grid. Goggles. Dropper bottles of .
Magnesium nitrate $Mg(NO_3)_2$ potassium carbonate K_2CO_3 .
Calcium nitrate $Ca(NO_3)_2$ potassium sulfate K_2SO_4 .
Strontium nitrate $Sr(NO_3)_2$ potassium chromate K_2CrO_4

Alkaline Earth Metals Lab - olatheschools

Calculate the K_{sp} for this alkaline earth hydroxide. Expert
Answer Balanced chemical equation is: $HCl + OH^- \rightarrow H_2O + Cl^-$
Here: $M(HCl)=0.143\text{ M}$ $V(HCl)=25.9\text{ mL}$ $V(OH^-)=27.5\text{ mL}$
view the full answer

Solved: An Alkaline Earth Hydroxide, $M(OH)_2$, Was Taken To

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The experiments with the alkaline earths involve determining the relative solubilities of the salts formed by the alkaline earth cations with sulfate, carbonate, oxalate and chromate ions. When solutions containing these M^{+2} cations are mixed with the above X^{-2} anions, the following reaction will occur if the salt MX is not very soluble: $M^{+2} (aq) + X^{-2}$

Lab #4 The Behavior of Two Families in the Periodic Table
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Alkaline Earth Metals Lab Answer Key

In this lab, we investigate the reaction of two alkaline earth metals, magnesium and calcium, and a group 3 metal, aluminum, with water.

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