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$\times 10^2$ N 2. 64.9 kg

3. 1.7 N/kg Page 80

Quick Check 1a. 1 4

F g 1b. 1 9 F g 1c.

g 4F 1d. g 9F 2a.

9.80×10^2 N 2b.

1.62×10^2 N 2c.

1.4×10^{-11} N

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Questions 1. 600 N

2. F m 3m =

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3F.3a. $7.8 \times 10^2 \text{ N}$

3b. $1.4 \times 10^2 \text{ N}$ 4.

80.5 kg 5. 3.61

m/s²; 37% 6. 143.9

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Sulfur-1 6 7

magnesium-24

magnesium-26 12,

(a) 32 g Mass

Number 22 29 24

26 Atomic Number

10 14 12 12

Number of

Neutrons 12 12 14

13. ...

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destroyed, or
divided into smaller
particles. 3. All
atoms of the same
element are
identical in mass
and size, but they
are different in
mass and size from
the atoms of other
elements. 4.

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