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Answers Activity A

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A 40.0 L tank of ammonia has a pressure of 12.7 kPa. Calculate the volume of the ammonia if its pressure is changed to 8.4 kPa while its temperature remains constant.

Charles' Law Problems: $V_1 T_2 = V_2 T_1$. 1 atm = 760.0 mm Hg = 101.3 kPa. Calculate the decrease in temperature when 6.00 L at 20.0 °C is compressed to 4.00 L.

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