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Heat, Calorimetry, and Enthalpy. Make the following conversions: 444 cal to joules. 1.8 kJ to joules. 0.45 kJ to calories.

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SECTION 17.1 THE
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(pages 505-510) This

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section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. ~ Energy

Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that occur during chemical reactions?

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Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

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(pages 505—510)~~

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$$Q = m C \Delta T \quad H = -q$$

$$C = 4184 \text{ J moles } ^\circ\text{C}^{-1}$$

When 150-g sample of KCl dissolves in 650 g of water in a calorimeter, the temperature drops from 310°C to 205°C

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thermochemistry. The
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TRO For this chapter we

will review some of

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Thermochemistry, and

study a part of Chapter 6

which we omitted in

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General Chemistry I.
Everything in the
universe is energy or
matter.

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