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venturinafrench. 1.1 The

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Thinking Like a Scientist

1.4 Problem Solving in
Chemistry. Terms in this set
(25) matter. Anything that
has mass and takes up space.

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Chemistry, like all sciences, is quantitative. It deals with quantities, things that have amounts and units. Dealing with quantities is very important in chemistry, as is relating

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quantities to each other. In this chapter, we will discuss how we deal with numbers and units, including how they are combined and manipulated.

Chapter 1 - Measurements -

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**CHE 105/110 - Introduction
to ...**

Introduction to Chemistry. 2
Burning log Rusting nail.
BIGIdeaChemistry is a
science that is central to
our lives. 1.1 A Story of
Two Substances.

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MAINIdeaChemistry is the study of everything around us. 1.2 Chemistry and Matter. MAINIdeaBranches of chemistry involve the study of different kinds of matter. 1.3 Scientific Methods.

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Chapter 1: Introduction to Chemistry

Chapter 1: Introduction to
Chemistry. Jennie L.
Borders. Section 1.1 -
Chemistry. Matter is
anything that has mass and

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occupies space. Chemistry is the study of the composition of matter and the changes that matter undergoes. Branches of Chemistry.

Chapter 1: Introduction to Chemistry

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2 Chapter 1 Introduction to
Chemistry CHAPTER 1 Visit
the Chemistry Web site at
science.glencoe.com to find
links about chemistry and
matter. The four nebulae
shown here contain a stew of
elements. The red color in

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two of the nebulae is emitted by hydrogen atoms. The Horsehead Nebula can be seen on the right. The fourth nebula is the bluish structure below the horse's head.

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What ...

Chapter 1 - An Introduction
to Chemistry 5 Exercises Key
Exercise 1.1 - Units Derived
from Metric Prefixes:
Complete the following

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Relationships. Rewrite the relationships using abbreviations for the units.

(Obj 7) a. 1 megagram = 10^6 gram
b. 1 milliliter = 10^{-3} liter
Exercise 1.2 -

Uncertainty: If you are given the following values

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Chemistry" Tools. Copy this
to my account; E-mail to a
friend; Find other

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Worksheet Answers;
Start over;
Help; Check your knowledge
of the vocabulary presented
in this chapter. A B; pure
chemistry: the pursuit of
chemical knowledge for its
own sake: technology: the
means by which a society

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Quia - Chapter 1

"Introduction to Chemistry"

1.1: The Scope of Chemistry

Chemistry is the study of matter and the ways in which

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different forms of matter combine with each other. You study chemistry because it helps you to understand the world around you. Everything you touch or taste or smell is a chemical, and the interactions of these

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chemicals with each other
define our universe.

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with a fresh theme that will
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CHAPTER 1: Introduction to
Chemistry. 1.1 The Nature of
Chemistry 1.2 Matter. Matter
- has mass, occupies space
Mass: measure of the
quantity of matter Extensive
property - depends on
quantity, i.e., mass, volume

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Intensive property – depends on identity of substance, i.e., density, color

Physical property - can be measured without changing the composition of the substance
Physical change - change in the form of the

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substance, but not its
chemical composition.

CHAPTER 1: Introduction to Chemistry

Chapter 1 - Introduction to
Chemistry - 1.2 Chemistry
and You - 1.2 Lesson Check -

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can be useful in 1)
explaining the natural
world, 2) preparing people
for career opportunities,
and 3) producing informed
citizens.

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Chapter 1 : Introduction to
Chemistry. Chapter 1
Vocabulary and Questions for
Chemistry. STUDY. PLAY. Why
is the scope of Chemistry so

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Chemistry affects all aspects of life and most natural events because all living and nonliving things are made of matter. Matter. Anything that has mass and takes up space.

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Mole concept numericals

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Chapter 1 An Introduction to Chemistry. For the Celsius scale, the temperature at which water freezes is defined as $0\text{ }^{\circ}\text{C}$, and the temperature at which water boils is defined as $100\text{ }^{\circ}\text{C}$. Thus a degree Celsius, $^{\circ}\text{C}$,

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is 1/100 of the temperature difference between freezing and boiling water (Figure 1.9).

**Chapter a I to ChemIstry -
An Introduction to Chemistry**

Chapter 1 Notes:

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Introduction to Chemistry I.

What is Chemistry? Chemistry is the study of matter and the changes that matter undergoes. Matter = Anything that has mass and occupies space. Examples of Matter? Air, Water, A desk, YOU!

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Examples of Non-matter?

Heat, Light, Sound, Energy

3. II. Five Major Branches

of Chemistry1. Organic

Chemistry - study of

essentially all chemicals

containing carbon2.

Inorganic Chemistry - study

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of chemicals, in general,
that do NOT contain carbon³.

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introduce this chapter's Big
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discussion of chemistry that occurs around them every day. Ask students to give an example of chemistry that is occurring in the classroom not related to lab experiments. Accept all reasonable responses.

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