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(and why is now
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Pant 16 | For

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~~Receives Highest~~

~~Dose of Nuclear~~

~~Radiation—This Is~~

~~What Happened To~~

~~Him Speed of Light,~~

~~Frequency, and~~

~~Wavelength~~

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the rays and

particles emitted

by the radioactive

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source.

radioisotope.

unstable isotopes

(any element past

Pb)

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ISBN-10:

0132525763,

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Transformations

25.3 Fission and

Fusion 25.4

Radiation in Your

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Life... Sample

Problem 25.1

Carbon-14 emits

beta radiation and

decays with a half-

life ($t_{1/2}$) of 5730

years. Assume that

you start with a

mass of $2.00 \times$

10^{-12} g of

carbon-14. 1 2 a.

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radioactive decay,
is an example of a.

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nuclear reaction. □

Nuclear reactions begin with

unstable. isotopes, or radioisotopes. □

Atoms of these isotopes become more stable. when changes occur in their nuclei. □ The changes are always accompanied by the. emission of large amounts of

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energy. 25.1

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treaty can be
gotten by just
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nuclear radiation
answers afterward
it is not directly
done, you could
put up with even

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life, in the region of
the world.

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Types of Radiation

Discuss Explain

that the nuclei of a
radioactive
element

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spontaneously
decompose.

Nuclear chemistry is the study of changes in matter that originate in atomic nuclei. Ask, What types of radiation exist, and how harmful are they? (The three most common types of radiation emitted by

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unstable nuclei are

Radiation

25.1 Nuclear

Radiation 25

Mass number and atomic number are conserved in all nuclear reactions.

The mass of a 25.0 g piece of ^{238}Cm (half-life: 2.4 hr)

will be reduced to 147 3.1 g after 7.2 hr. Study Guide for

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The two general kinds of nuclear reactions are nuclear decay reactions and nuclear transmutation reactions. In a nuclear decay reaction, also called radioactive

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decay, an unstable nucleus emits radiation and is transformed into the nucleus of one or more other elements. The resulting daughter nuclei have a lower mass and are lower in energy (more stable) than the parent nucleus that decayed.

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