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(1-3) Chapter 9  
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9.1 Introduction to  
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*Ideal Stoichiometric  
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~~Experiment #7: The  
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Chemical Reaction.~~

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to Grams (using a  
balanced equation)*

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*to Stoichiometry*

*Chapter 9*

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*CHEM 103 - Chapter 9*

*Chemical Equation*

*Calculations (aka*

*Stoichiometry) Part 1*

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Stoichiometry Basic

Introduction, Mole to

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Mole, Grams to Grams,  
Mole Ratio Practice  
Problems CH Ideal

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Mr C *CHEM 103* -

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Equation Calculations  
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*Part 2 Rules - Chapter 9*

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**Example 4 Heat of**

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deals with elements in

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reactants and products  
in chemical reactions,  
focusing on.

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Stoichiometry MIXED

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ANSWER Answer the

following questions in

the space provided. 1.

Given the following

equation:  $C_3H_4(g) +$

$xO_2(g) \rightarrow 3CO_2(g) +$

$2H_2O(g)$  4 a. What is

the value of the

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coefficient x in this equation? 40.07 g/mol

b. What is the molar mass of  $C_3H_4O_2$ ? 2 mol  $H_2O$  c. What is the mole ratio of  $O_2$  to  $H_2$

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*Page 13/34*

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chemical equations and  
the law of conservation  
of mass. All reaction  
stoichiometry

calculations start with a  
balanced chemical  
equation. This equation  
gives the

CorrectionKey=NL-A

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the books Page 10/29

## Stoichiometry

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Chapter 9 focuses on reaction stoichiometry: using a balanced chemical equation to calculate the number of grams, moles, or particles of reactants/products involved in a chemical

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reaction. Students had an introduction to composition stoichiometry in Chapter 3 and will now move on to some more difficult problems.

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high sprit may not feel

bored or give up

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Answer Key

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ANSWER Answer the

following questions in  
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Given the following  
equation:  $C_3H_4(g) + x.$

$O_2(g) ( 3CO_2(g) +$

$2H_2O(g)$  a. What is the  
value of the coefficient .

x. in this equation? b.

What is the molar mass  
of  $C_3H_4$ ? c. How many

moles are in an 8.0 g

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sample of  $C_3H_4$ ?

Stoichiometry

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Answers Chapter 12

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squarely in front of

these Hebrews. Those

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who doubt Jesus have  
sinned willfully,  
trampled under foot the  
Son of God, regarded as  
unclean the blood of  
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SECTION 12.1 THE

ARITHMETIC OF

EQUATIONS (pages

353–358) This section

explains how to

calculate the amount of

reactants required or



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product formed in a  
nonchemical process.

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#### Reading Study Work

The American Pageant

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Questions 1. What is  
significant about how  
Massachusetts drafts its  
constitution? It called a  
special convention to

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draft its constitution and then submitted the final draft directly to the people for ratification 2. What things did the new state constitutions have in common?

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