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## Chemical Engineering Formula Sheet

Basic Thermodynamic Formulas (Exam Equation Sheet) Control Mass (no mass flow across system boundaries) Conservation of mass:  $\dot{m} = \dot{m}_{in} - \dot{m}_{out}$ . Conservation of energy (1st Law):  $\dot{Q} - \dot{W} = \dot{m}(h_2 - h_1) + \dot{m} \frac{V_2^2 - V_1^2}{2} + \dot{m} g(z_2 - z_1)$ .

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reference? you can download the Pocket Guide to Chemical Engineering on this link: [HERE](#) The book is written by Carl R. Branam who has more than 40 years of experience in process design and troubleshooting in oil, chemicals, fibers, polymers, coal, natural gas, and LNG.

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Equations and Formulas of Chemical Reaction Engineering  
Type or Name General Formula 1. Alkane or paraffin (also saturated hydrocarbons) 2. Alkene or olefin (unsaturated hydrocarbons) 3. Alkyne 4. Alcohol 5. Ether 6. Aldehyde 7. Ketone 8. Carboxylic Acid 9. Grignard reagent 10. Acyl halide 7 13282AICEtext 4/12/04 12:20 PM Page 7

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iii PREFACE The Fundamentals of Engineering (FE) Supplied-Reference Handbook is the only reference material allowed in the FE examination. Many examinees find that it is helpful to review

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this book before exam day to become familiar with the reference material it contains.

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Chemical Engineering Formula Sheet where reaction is in solution, ([ ] refers to molarity) where reaction is in the gas phase, (p partial pressure) Gram equivalent weight A. (nonredox reaction) the mass in grams of a substance equivalent to 1 gram-atom of hydrogen, 0.5 gram- atom of oxygen, or 1 gram-ion of the hydroxyl ion.

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### NCEES FE exam information

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This list gives you some of the most common conversion factors you need in thermodynamics. Acceleration:  $1 \text{ m/s}^2 = 100 \text{ cm/s}^2$ . Area:  $1 \text{ m}^2 = 10^4 \text{ cm}^2 = 10^6 \text{ mm}^2$ . Density:  $1 \text{ g/cm}^3 = 1 \text{ kg/L} = 1,000 \text{ kg/m}^3$ . Energy, heat, work, internal energy, enthalpy:  $1 \text{ kJ} = 1,000 \text{ J} = 1,000 \text{ N}\cdot\text{m} = 1 \text{ kPa}\cdot\text{m}^3$ .  $1 \text{ kJ/kg} = 1,000 \text{ m}^2/\text{s}^2$ .

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