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Monoisocyanates 1 1.1

Alkyl and Aryl

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Isocyanates are used in the form of methylene diisocyanate (MDI) and polymethylene polyphenyl diisocyanate (PMDI), with rigid foam applications being the largest consumer of MDIs in the recent past (Anonymous, 2015b).

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Isocyanates, upon the action of water, release CO<sub>2</sub> that facilitates the formation of a microcellular foam. In addition, as a polyfunctional reactant, PMDI enables the polyol linkages through interaction with hydroxyl groups, gradually establishing a cross-linked polymer ...

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Isocyanate reactions are very susceptible to catalysis that is catalysts greatly increase the rate of nucleophilic addition of compounds with active hydrogen to the C=N bond. Widely used catalysts include tertiary amines, such as triethylene diamine

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(TEDA),  
dimethylpiperazine  
(DMP),  
dimethylethanolamine  
(DMEA), and  
diazabicyclooctane  
(DABCO), as well as tin  
compounds like  
dibutyltin dilaurate  
(DBTL).

### Isocyanate Reactions

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highlights the industrial  
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both mono- and  
diisocyanates is of ever  
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the chemical industry.  
The annual global  
consumption of  
polyurethanes, derived  
from diisocyanates, has  
reached over six million  
metric tons.

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The most common  
forms of isocyanates are

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toluene diisocyanate (TDI), methylene diphenyl diisocyanate (MDI) and hexamethylene diisocyanate (HDI). TDI is a liquid at room temperature, and can cause asthma-like conditions when inhaled as an aerosol. TDI is a key ingredient in many spray paints and coatings.

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