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SHORT ANSWER

Answer the

following questions

in the space

provided. 1. Solid

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CaCl_2 does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl_2 is an ionic solid. In the crystal form, its ions are locked in position.
Dissolving

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SHORT ANSWER 1.

c 2. a 3. b 2. a.

alcohol b. water c.

the gels 3. The

mixture is a colloid.

The properties are

consistent with

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those reported in
Table 3 on page
404 of the text.

The particle size is
small, but not too
small, and the
mixture

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