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and Its

Properties Matter

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– Anything that has MASS and VOLUME. ATOMS and MOLECULES are the basic building blocks of matter.

ATOM-- the smallest unit of an element that still maintains the properties that element.

ELEMENT– a pure

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Substance made
up of only one
type of atom.

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in Column A with
the term in
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A Column B 1.

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frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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property.

intensive

property.

extensive

properties

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intensive

property

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property that depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not... mass, volume, length, weight.

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11 31 number of

moles of known

substance C H C

H C H O O O CO

CO CO H O H O H

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0 number of
moles of unknown
substance 0 C O H
0 C H C O H 0 C H
0 H 0 C H 0 C O
use mole ratio 5
mol 0 2 1 mol C
3 H 8 3 mol C O 2
1 mol C 3 H 8 4
mol H 2 0 1 mol
C 3 H 8 1 mol C
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A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which one or more substances are converted

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into different substances. Heat and light are often evidence of a chemical change.

Properties of Compounds Quite different properties than their component elements.

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