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An ideal gas is a hypothetical gas consisting of identical particles with zero volume and with no intermolecular forces. Additionally, the constituent atoms or molecules undergo perfectly elastic collisions with the walls of the container. An ideal gas can consist of molecules (e.g. carbon dioxide molecules, CO₂) or atoms (e.g. neon atoms, Ne). Real gases do not exhibit these exact prop-

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Plants convert carbon dioxide, water, and energy from the sun into sugar ($C_6H_{12}O_6$) and dioxygen in the process of photosynthesis.

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