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The AP Chemistry course is taught as a continuation of the first year course. As such, units are numbered accordingly. Units 1-8 are covered during the first year course and reviewed over a short time span during the AP course. (Links for these units can be found under the "CHEMISTRY" tab above and on the links immediately below this overview.)

## Units 1-5 Pre-AP Chemistry Review - Mrs. Forest's ...

Chemistry Unit Review Sheet- Part 2. Read the descriptions below and determine whether each item describes an endothermic or an exothermic reaction. \_\_\_\_\_ 1. A reaction that produces heat energy \_\_\_\_\_ 2. A reaction that absorbs energy \_\_\_\_\_ 3. Paper burns \_\_\_\_\_ 4

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Unit 5 Review. Show all questions <= ... 2.5 moles per second; Given the following reaction:  $C_3H_8(g) + 5O_2(g) \rightarrow 3CO_2 + 4H_2O(g)$  If the initial rate of the reaction produces 2 moles of water vapor,  $H_2O$ , per second, what is the rate at which propane,  $C_3H_8$ , is consumed initially? ?

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36.5 g H M = moles of solute liters of solution 0.200 M = mol solute 0.250 L mol solute = 0.0500 mol HCl? g HCl = 0.0500 m HCl ! Cl 1 mol HCl = Cl 4. A solution of Na 2CO 3 contains 65.0 g of solute dissolved in water to make a 3.00 M solution. What is the volume of the solution, in liters? ? g HCl = 0.0500 mol HCl ! 36.5 g HCl 1 mol HCl = 1.83 ...

### 10-24a.b.c.d Unit 10 Review Sheet-Key

Chemistry 12 Unit 1 - Reaction Kinetics Unit 1 - Review Sheet Page 2 2. For each of the following reactions find a quantity or property which could be monitored in order to measure the rate of reaction See p. 2-5 in SW. ("a" is done as an example.) a)  $3\text{H}_2(\text{g}) + \text{N}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$  pressure will decrease as reaction proceeds because

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