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Answers

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Definitions of Acids and Bases

Arrhenius definitions 1) acid:

substance that when dissolved in water increases $[H^+]$; (note: H^+ exists bonded to water as the

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hydronium ion, H_3O^+ , so $[\text{H}^+]$ and $[\text{H}_3\text{O}^+]$ are equivalent expressions)

2) base: substance that when dissolved in water increases $[\text{OH}^-]$

Bronsted-Lowry definitions ...

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exactly it measures. The pH is a measure of the concentration of hydrogen ions (H^+) in a solution. Recall that in water H^+ ions bond to water to form H_3O^+ (hydronium ions). Keep ...

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FIGURE 1: Bohr Diagrams of
boron,carbon and nitrogen nucleus

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charge = +6 nucleus charge This PDF book provide chemquest 49 guide. To download free chemquest 16 you need to register. ChemQuest #1 Answers. ChemQuest 1. Name: . 49. $(5.9 \times 10^9) \times (6.3 \times 10^{-5}) = (5.9)(6.3) \times 10^{9+(-5)} = 37.17 \times 10^4 = 3.717 \times 10^5$. 5. When you divide two numbers, .

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