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Electroplating is a general name for processes that create a metal coating on a solid substrate through the reduction of cations of that metal by means of a direct electric current. The part to be coated acts as the cathode of an electrolytic cell; the electrolyte is a solution of a salt of the metal to be coated; and the anode is usually either a block of that metal, or of some inert conductive material. The current is provided by an external power supply. Electroplating is widely used in indus

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Metal finishing has transformed itself within a few decades from what was once an empirical craft into a key technology grounded on scientific principles. 1. Modern electroplating is a form of metal finishing used in various industries, including aerospace, automotive, military, medical, RF microwave, space, electronics

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and battery manufacturing. It is the electrochemical process whereby metal ions in solution are bonded to a metal substrate via electrodeposition.

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Electroplating is the application of electrolytic cells in which a thin layer of metal is deposited onto an electrically conductive surface. A cell consists of two electrodes (conductors), usually made of metal, which are held apart from one another. The electrodes are immersed in an electrolyte (a solution).

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Electroplating is the process of plating one metal onto another by hydrolysis, most commonly for decorative purposes or to prevent corrosion of a metal. There are also specific types of electroplating such as copper plating, silver plating, and chromium plating.

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