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Activity Basic

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Activity Page 2 Basic

Stoichiometry Model 2

$2A + 3B \rightarrow 5C + 4D$ If 3

mol A react, how many

mol C are produced? 3

mol A * 5 mol C = 7.5

mol C 2 mol A 5. If 21.0

mol A are reacted, how

many mol C are

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produced? Using
dimensional analysis,
show how you
calculated your answer.

6.

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Activity Topic:

Measurement: Scientific

Mathematics. Unit

Dimensional Analysis

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Activity 1 Version 2.

Why? In this activity we will see that it is possible to look at a situation from several points of view, or . to take measurements of that same situation using different units of measure. Every measurement has 2 . components: magnitude

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Activity «Activity Pogil

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Activity Basic

Stoichiometry. Basic

Stoichiometry - KEY.

Why? In this activity we will address the

question: How do I

convert between

different chemical

species in a given

reaction? Model 1. $2A +$

$3B \rightarrow 5C + 4D$. 2 mol A

produces 5 mol C 4 mol

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A produces 10 mol C. 3
mol B produces 4 mol D
6 mol B produces 8 mol
D

Answers

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Activity - Science Done

Wright

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Activity Page 12 Basic

Stoichiometry Model 3

Given the following

equation: $\underline{\quad} 1 \underline{\quad} \text{N}_2 (\text{g})$

$+ \underline{\quad} 3 \underline{\quad} \text{H}_2 (\text{g}) \underline{\quad} 2 \underline{\quad}$

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NH_3 (g) 25. If 0.052 mol N_2 are reacted, how many mol NH_3 are formed? Using dimensional analysis, show how you calculated your answer.

$$0.052 \text{ mol N}_2 * 2 \text{ mol NH}_3 / 1 \text{ mol N}_2 = 0.104 \text{ mol NH}_3$$

Point out the important parts of their definitions and ...

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Activity Page 5 Basic

Stoichiometry _____ H

$2 (g) + \underline{\hspace{1cm}} O_2 (g)$

$\underline{\hspace{1cm}} H_2O (g)$ 14.

Given the equation above, determine the number of moles of water produced when 5.2 g O_2 are reacted. Make sure to show the dimensional analysis in your work. View full document.

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1(BW) Particle
connections □ What's in
a name? Why? In this
activity we will address
the question: How do
the smallest particles of

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matter connect to . each
other and how do we
represent those
connections by the
names we give a
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Activity Topic:

Stoichiometry. Basic
Stoichiometry. Why? In
this activity we will
address the question:

How do I convert
between different
chemical species in a
given reaction? Model
1. $2A + 3B \rightarrow 5C + 4D$. 2

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mol A produces 5 mol C
4 mol A produces 10
mol C. 3 mol B
produces 4 mol D 6 mol
B produces 8 mol D

HS Chemistry POGIL
Activity

6 POGIL Activities for
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20. For each experiment
in Model 2, determine
the relationship between
the independent and

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dependent variables, and write an algebraic expression for the relationship using variables that relate to those in the experiment (P internal, V, T or n). Use k as a proportionality constant in each equation.

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experience and triumph

by spending more cash.

still when? attain you

how to that you require

to acquire those all

needs as soon as having

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Activity Name: Date:

Basic Stoichiometry

Why? In this activity we
will address the

question: How do I
convert between moles
of different chemical
species in a given

reaction? Model 1 $2A +$

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3B 5C + 4D 2 mol A
produces 5 mol C 4 mol
A produces 10 mol C 3
mol B produces 4 mol D
6 mol B produces 8 mol
D ...

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