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Livingston helps make the complex concepts behind the electronic properties of materials much more accessible for students. His very readable writing style and clear organization help to make the key topics much easier to understand. The first part of this text presents only [classical] ideas, covering the electronic properties of solids that are pertinent to the use of materials as components in various products.

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James Livingston has written a highly readable undergraduate text introducing the physics and chemistry underlying the electronic Living properties of engineering solids. The first half of the text uses a semi-classical approach, while the second half introduces quantum mechanics and applies quantum chemistry and quantum physics to the basic properties of metals, insulators, and semiconductors.

Electronic Properties of Engineering Materials | Wiley

This text on the electrical, optical, magnetic, and thermal properties of materials stresses concepts rather than mathematical formalism. Suitable for advanced undergraduates, it is intended for materials and electrical engineers who want to gain a fundamental understanding of alloys, semiconductor

Electronic Properties of Materials | Rolf E. Hummel | Springer

To finalize the material for an engineering product / application, we should have the knowledge of Electrical properties of materials. The Electrical properties of a material are those which determine ability of material to be suitable for a particular Electrical Engineering Application. Some of the typical Electrical properties of engineering

Electrical Properties of Engineering Materials | Electrical4U

The effect of water exposure on amorphous indium-gallium-zinc oxide (a-IGZO) semiconductors was reported. It was found that water can diffuse in and out of the a-IGZO film, reversibly affecting the transistor properties. Two competing mechanisms depending on the thickness of the active channel were clarified.

Electronic transport properties of amorphous indium

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An electrolyte solution is a solution that generally contains ions, atoms or molecules that have lost or gained electrons, and is electrically conductive. For this reason they are often called ionic solutions, however there are some cases where the electrolytes are not ions. For this discussion we will only consider solutions of ions.

Electrolyte Solutions - Chemistry LibreTexts

Carbon nanotube material can now be produced in macroscopic quantities. However, the raw material has a disordered structure, which restricts investigations of both the properties and applications of the nanotubes. A method has been developed to produce thin films of aligned carbon nanotubes. The tubes can be aligned either parallel or perpendicular to the surface, as verified by scanning ...