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When there are no external forces, the shape of a liquid drop is determined by. (A) Surface Tension of the liquid. (B) The density of the liquid. (C) The viscosity of the liquid. (D) The temperature of air only.

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Surface tension is the tendency of fluid surfaces to shrink into the minimum surface area possible. Intermolecular forces such as Van der Waals Force draw the liquid particles together.

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MCQ on Surface Tension. 1. When there are no external forces, the shape of a liquid drop is determined by (a) Surface Tension of the liquid (b) The density of the liquid (c) The viscosity of the liquid (d) The temperature of air only. Answer. Answer: (a) Surface Tension of the liquid

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PHYSICS NOTES SURFACE TENSION & VISCOSITY

Surface Tension Multiple Choice Questions and Answers. Surface Tension MCQ Questions and Answers Quiz. 31. Which is the effect of increase in the temperature on the surface tension of the liquid? it decreases; it increases ; it remains the same; None of the above

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Mcqs About Surface Tension With Answers

Physics GK 11 - Surface Tension & Capillarity Physics Multiple Choice Questions (MCQs) and Answers with explanation on Surface Tension & Capillarity for SSC, PCS, UPSC, IAS, NTSE, CLAT, Railways, NDA, CDS, Judiciary and other examinations of India.

Physics GK 11 - Surface Tension & Capillarity - GKToday

Q3: The surface tension of a liquid is 70 dyne/cm. In MKS system its value is? (a) 70 N/m (b) 7×10^{-2} N/m (c) 7×10^2 N/m (d) 7×10^3 N/m Answer. b) 7×10^{-2} N/m. Q4: The dimensions of Kinetic energy is same as that of (a) Force (b) Pressure (c) Work (d) Momentum Answer. c) Work. Q5: At 4° C, the density of water is equal to (a) 10^{-3} kg m⁻³

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a) Reduction in the interfacial tension between the phases b) Increase in the interfacial tension between the phases c) Catalysation of a chemical reaction between the phases

Chapter 9: Multiple choice questions - Oxford University Press

Water rises in a capillary upto a extension height such that upward force of surface tension balances the force of 75×10^{-4} N due to weight of water. If surface tension of water is 6×10^{-2} N/m. The internal circumference of the capillary must be. 12.5×10^{-2} m. 6.5×10^{-2} m. 0.50×10^{-2} m. 1.25×10^{-2} m

Important Questions of Mechanical Properties of Fluids for ...

Surface tension of a liquid is due to force of attraction between like molecules of a liquid i.e., cohesive force between molecules. Surface tension may be defined as _____. the work done per unit area in increasing the surface area of a liquid under isothermal conditions.

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