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SECTION 3.1 MEASUREMENTS AND THEIR UNCERTAINTY (pages 63 – 72)

This section describes the concepts of accuracy, precision, and error in measurements. It also explains the proper use of significant figures in measurements and calculations.

~~SECTION 3.1 MEASUREMENTS AND THEIR UNCERTAINTY~~

Measurements and Their Uncertainty

OBJECTIVES: – Distinguish among accuracy, precision, and error of a measurement. 4 Section 3.1 Measurements and Their Uncertainty OBJECTIVES:

– Determine the number of significant figures in a measurement and in a calculated answer. 5 Measurements Qualitative measurements are words, such as heavy or hot

~~Chapter 3 Measurements and Their~~

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When a measurement reported as 5.0 kg is divided by 3.0 L, for example, the display may show 1.666666667 as the answer. We are justified in reporting the answer to only two significant figures, giving 1.7 kg/L as the answer, with the last digit understood to have some uncertainty.

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20. Give the Of significant figures in the following measurements, 3.85 x b. 17.30 cmS perform the operations and give in Standard form With the number of significant 37.2mL. 8.382 Ck. b. a.1Z9Y10 = 2.1

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accuracy, Precision, and errorAccuracy, precision, and error help determine the reliability of measurements. The accuracy of a measurement is determined by how close the measured value is to the actual value. The precision of a measurement is determined by how close repeated measurements are to one another.

Error is the difference between the measured value and the accepted value.

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The uncertainty can also be stated as a percentage of the measured value. This is called the percentage uncertainty, $\%$. This can be calculated by taking the absolute uncertainty and dividing it by the mean, or measured value as below.
$$\text{percentage uncertainty} = \frac{\text{absolute uncertainty}}{\text{mean value}} \times 100\%$$

~~Mr Toogood Physics – Measurements and
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