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Stoichiometry Chapter Review Answers 1.

2. Mole Ratios a. Mole ratios are
conversion factors that relate the number
of moles of one chemical to 0.0209 mol

AgNO₃ Aspirin production a. 13.5 kg

C₉H₈O₄ b. 7.66 kg C₄H₆O₃ c. 4.29 L

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HC2H3O2 Ideal stoichiometry
calculations do not account for...

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PROBLEMS General Stoichiometry Do
not assume that equations without listed
coefficients are balanced. 9. Given the
chemical equation $\text{Na}_2\text{CO}_3(\text{aq}) +$

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$\text{Ca(OH)}_2(\text{s}) + 2\text{NaOH}(\text{aq}) + \text{CaCO}_3(\text{s})$,
determine to two decimal places the molar
masses of all substances involved, and
then write them as conversion factors.

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Stoichiometry Stoichiometry comes from the Greek words stoicheion, meaning "element The number of significant figures in the answer is therefore determined only by the number of 10 oxygen molecules 5 carbon dioxide molecules 5 oxygen molecules in EXCESS CHAPTER 9...

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Chemistry (4th Edition) Burdge, Julia
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