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| exercise solution part-1
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Chapter 10 -

**Chemistry Class 12th -
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ORganic Chemistry

????? ??? ????? ??? ?

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Organic Chemistry I

Simple Trick to

Understand Conversion

Reactions Of Organic

Compounds ~~ORGANIC~~

~~CONVERSIONS~~

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~~U0026 HALOARENE~~

Chemical properties of
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(NCERT) Chemistry-

~~HALOALKANES AND~~

~~HALOARENES |~~

~~CHAPTER 10 |~~

~~CLASS 12 | Pathshala (~~

~~hindi) #1 Haloalkane~~

~~and Haloarene (inter~~

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CHEMISTRY

Haloalkanes

Haloarene [Intext

Questions 10.1 to 10.5]

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*Class 12th Exercise 10-
Ques No.10.21 6.*

*Reactions of
Haloalkanes Ncert*

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Chemistry NCERT IN

TEXT QUESTIONS.

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10.1 Write structures of the following compounds: (i) 2-Chloro-3-methylpentane (ii) 1-Chloro-4-ethylcyclohexane

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~~12 Chemistry Chapter~~
~~10 Haloalkanes ...~~

Ans. NCERT Solutions

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Haloalkanes and

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uses, and physical and

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solutions of Haloalkanes
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Chemistry explained in

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detail by experts to help students prepare for their CBSE exams.

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Haloalkanes can be classified based on the carbon atom to which the halogen atom is attached. When a halogen atom is connected to a primary,

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secondary, or tertiary carbon atom, it is referred to as a primary, secondary, or tertiary haloalkane, respectively.

Example: $\text{CH}_3\text{CH}_2\text{CH}_2\text{Cl}$

1-chloropropane (1°)

~~Haloalkanes and
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detailed explanation that
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10th chapter in the
NCERT 12 Chemistry
Solutions textbook deals
with haloarenes and
haloalkanes. Once the
chapter is completed,

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students will be able to name haloarenes and haloalkanes based in the IUPAC nomenclature from the given structure.

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Chemistry Chapter 10 –
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Chapter 10 Haloalkanes
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Haloalkanes and
Haloarenes is an
important chapter from
examination
perspective. This
chapter defines
haloalkanes and
haloarenes according to

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the IUPAC system of nomenclature. The reactions involved in the preparation of haloalkanes and haloarenes are also explained here.

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Since the cycloalkane reacts with Cl_2 in the presence of bright

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sunlight to give a single monochloro compound, C_5H_9Cl , therefore, all the ten hydrogen atoms of the cycloalkanes must be equivalent. Thus, the cycloalkane is cyclopentane.

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Patrick Pringle was born in 1935 in Rochester, New York. He was a freelance writer and photographer, wildlife biologist and educator. He was the editor, executive editor of Nature and Science

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Chem during 1965 – 1967.

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kCN is predominantly

ionic and provides

cyanide ions in solution,

which is ambident

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nucleophile and bind with carbon side to form as the major product, while AgCN is covalent and form isocyanide as the major product.

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