

## Nuclear Chemistry Section Review Answer Key

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~~Nuclear Chemistry Section Review Answer~~  
Base your answers to questions 33 and 34 on the information below. Scientists are investigating the production of energy using hydrogen-2 nuclei (deuterons) and hydrogen-3 nuclei (tritons). The balanced equation below represents one nuclear reaction between two deuterons.  $2\ ^1_1\text{H} + 2\ ^1_1\text{H} \rightarrow 3\ ^2_2\text{He} + 1\ ^0_0\text{n} + 5.23 \times 10^{-13}\ \text{J}$  33. Identify the type of nuclear reaction represented by the equation.

~~unit 11- nuclear chemistry review packet key.pdf - 1 Base ...~~

~~CHAPTER 21 REVIEW Nuclear Chemistry SECTION 2 SHORT ANSWER~~ Answer the following questions in the space provided. 1. \_\_\_\_ The nuclear equation is an example of an equation that represents (a) alpha emission. (b) beta emission. (c) positron emission. (d) electron capture. 2. \_\_\_\_ When undergoes electron capture to form a new element X, which of

~~CHAPTER 21 REVIEW Nuclear Chemistry~~

~~Chapter 21 Review Nuclear Chemistry Answer Key Chemistry Chapter 21 Nuclear Chemistry Test Review. Flashcard maker : August Dunbar. nucleons. protons and neutrons. nuclide. An atom identified by the number of protons and neutrons in its nucleus. mass defect. The difference between the mass of an atom and the sum of the masses of its protons, neutrons, and electrons.~~

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