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Pearson Science Grade 7K - Chemical Building Blocks ...

Sample answer: An element is a pure substance that cannot be broken down into any other substances. A compound is a pure substance made of two or more elements chemically combined in a set ratio. 25. Sample answer: According to the law of conservation of mass, matter is not created or destroyed in any chemical or physical change.

Chapter 1 Test - Introduction to Matter

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Observe changes in physical properties of materials. For instance, formation of a precipitate, gas production, and the color change are possible evidence that a chemical reaction has taken place.

CH 5 Lesson 1 (Observing Chemical Change) Flashcards | Quizlet

Recognizing Chemical Changes (page 54) 7. List the four possible clues to a chemical change? 8. Is the following statement true or false? If you observe a clue for chemical change, you can be certain that a chemical change has taken place. ____ 9. Define a precipitate. Conservation of Mass (page 55) 10.

SECTION 2.1 PROPERTIES OF MATTER (pages 39-42)

Chemistry 12th Edition Chang, Raymond; Goldsby, Kenneth Publisher McGraw-Hill Education ISBN 978-0-07802-151-0

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2.4 Chemical Reactions and Enzymes Lesson Objectives Explain how chemical reactions affect chemical bonds. Describe how energy changes affect how easily a chemical reaction will occur. Explain why enzymes are important to living things. Lesson Summary Chemical Reactions Everything that happens in an organism is based on chemical reactions.

2.4 Chemical Reactions and Enzymes - Weebly

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a number in front of a chemical formula in an equation that indicates how many molecules or atoms of each reactant and product are involved in a reaction. synthesis a chemical reaction in which two or more simple substances combine to form a new, more complex substance.

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