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The separation of a mixture does not occur naturally or spontaneously.

Energy is required to separate the different molecular species. Exercise

1.3 Subject: Separation of a mixture requires a transfer of energy to it or

the degradation of its energy. Given: The first and second laws of

thermodynamics.

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Processes. • Absorption – Solutes removed from a gas into a liquid.

• Solutes removed from liquid into gas is called stripping or

desorption. • Distillation – Thermal vapor-liquid separation

processes (Ch 11); vapor phase generated from liquid • Liquid-liquid

extraction – Solute extracted from liquid A into an immiscible liquid

B (a solvent) • Leaching(extraction) – Solute extracted from a solid

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into a solvent phase ...

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