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 $\times 2.62 \times 10^{24}$ (b) # electrons
added = $Q e = 1.00 \times 10^{-3} \text{ C}$
 $1.60 \times 10^{-19} \text{ C electron} = 6.25$
 $\times 10^{15}$ or 2.38 electrons for
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mol 47.0 electrons atom =
 $\times 2.62 \times 10^{24}$ (b) # electrons
added = $Q/e = 1.00 \times 10^{-3} \text{ C}$
 $1.60 \times 10^{-19} \text{ C/electron} = 6.25$
 $\times 10^{15}$ or 2.38 electrons for
every 10^9 already present
23.2 (a) $F_e = k_e q_1 q_2 / r^2$
 $= 8.99 \times 10^9 \text{ N} \cdot \text{m}^2 / \text{C}^2$ 1.60
 $\times 10^{-19} \text{ C}^2 (3.80 \times 10^{-10} \text{ m})^2 =$
 $1.59 \times 10^{-9} \text{ N}$ (repulsion) (b)
 $F_g = G m_1 m_2$

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11cq; 11oq; 11p; 12cq; 12oq;
12p; 13cq; 13oq; 13p; 14cq;
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23.4 The Electric Field 23.5
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Ch. 23 - A particle with charge of 12.0 C is placed at the... Ch. 23 - A particle with charge $Q = 5.00$ C is located at... Ch. 23 - A particle with charge Q is located at the center... Ch. 23 - (a) A particle with charge q is located a distance... Ch. 23 - Find the net electric flux through (a) the closed...

~~A thin rod of length ℓ and uniform charge per unit length ...~~

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