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p = f(t) t+ O(t)2: (1.2)

Therefore, the contribution to the average electron momentum from the electrons that do not collide is p(t+ t) = (1 t ?)(p (t) + f) t+ O2: (1.3) Note that the contribution to the average momentum from

electrons which have collided will be of order (t)2and therefore negligible.

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It has one unpaired electron in the 3s orbital. The 3s valence atomic orbital of sodium overlaps with another such orbital of the same energy to form molecular orbitals. The

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Fe (s) + CuSO 4 (aq) −? FeSO 4 (aq) + Cu (s) Fe (s) + FeSO 4 (aq) −? no

reaction. Fe (s) + Al 2 (SO 4) 3 (aq) -? no reaction. Fe (s) + ZnSO 4 (aq) -? no reaction. No reaction occurs when any of the metals from zinc, iron, aluminium, and copper are placed in aqueous solution

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