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Entropy, Order Parameters, and Complexity James P. Sethna Laboratory of Atomic and Solid State Physics, Cornell University, Ithaca, NY 14853-2501 The author provides this version of this manuscript with the primary in-tention of making the text accessible electronically—through web searches

Entropy, Order Parameters, and Complexity

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Entropy in Solution Formation. For now, entropy can be thought of as molecular “disorder” or in terms of the energy of molecules and how spread out they are. This term increases with increasing temperature. As a molecule changes state, the general states of matter can be ordered as follows in terms of entropy: gases > liquids > solids.

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