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An electron is contained in a three-dimensional rectangular box given by $0 \leq x \leq a$, $0 \leq y \leq b$, and $0 \leq z \leq c$.
The solutions of the Schrödinger equation are specified by the quantum numbers n_x , n_y , and n_z . Recall
that the normalized wave function is...with energy...where n_x , n_y , and n_z are positive integers.

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r , and $l(l+1) \approx m^2 \approx 1/4$, which is to say, $l^2 + l + 1/4 = m^2$, or $(l + 1/2)^2 = m^2$, or $l = \sqrt{m^2 - 1/4}$. [Our present equation depends only on m^2 , and hence is the same for either sign, but the solution to Problem 4.39 assumed $l + 1/2 \approx 0$ (else u is not normalizable), so we need $|m|$ here.] Quoting 4.39: $E = (j$

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