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Chemistry Concepts and Applications Chapter 1: The Science of Matter Standardized Test Practice

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Chapter 11: Behavior of Gases Standardized Test Practice. Your Results: The correct answer for each question is indicated by a . 1: High on a mountain, the reading on a barometer decreases because _____. ... 13: 976 L of oxygen at STP is compressed into a 12.0-L tank at 273 K. What is the gas pressure in the

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tank?

Standardized Test Practice

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Chapter 2 Standardized Test Practice

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Chapter 6 Standardized Test Practice 1. The actual number of atoms of each element contained in a molecule is given by the molecule's: A. empirical

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formula. C. mass percent formula. B. molecular formula. D. percent composition. 2. What is the molar mass of the sodium sulfate Na_2SO_4 ? F. 71.06 g/mol
G. 94.05 g/mol H. 142.05 g/mol J. 215.13 g/mol 3.

Chapter 6 Standardized Test Practice

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