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WORKSHEET The Activity Series Use
with Chapter 9, Section 9.2 1. For each of
the following pairs of elements, underline
the one that would replace the other
element in a compound. Teaching
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Chapter 7, Section 7.2 1. How many
valence electrons does a neutral
magnesium (Mg) atom have? 2. What is
the charge on a magnesium ion? What

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Answer Key) does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl) atom have? 4.

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great conversation, but if you want to
improve your teaching, then transparency
makes a lot of sense.

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6. Considering your answer to question 5,
how many electrons can fit in the entire 5th
energy level? can (A_{Vel} is a ha AC5a)
50đ 7. Write the notation for an electron
spinning clockwise in a p sublevel in the
4th energy level.

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TEACHING TRANSPARENCY

WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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