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The elements carbon,

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hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things. They form the carbohydrates, nucleic acids, proteins, and lipids (all of which will be

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Key Chapter 24 defined later in this chapter) that are the fundamental molecular components of all organisms. In this chapter, we will discuss these important building blocks and learn how the unique properties of

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the atoms of different
elements affect their
interactions with other ...

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the different compounds
formed. Life revolves around
the balancing act between
the energy released as bonds
are broken and the energy
taken in as bonds are
formed. Life on earth
depends on the nature of the

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carbon atom and the nature
of water.

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For example, in Biology
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Life, students learn about
carbon, nitrogen, oxygen,
hydrogen, and iodine. The
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students the correct and incorrect ways of using the scales to solve problems and science units. However, as stated before, the online students will not be lectured about biology.

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Answers~~

SAMPLE ANSWER: Carbon is the primary element found in living things. SAMPLE

ANSWER: Organisms use carbon compounds to form four types

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of molecules: lipids,
carbohydrates, nucleic
acids, and proteins. SAMPLE
ANSWER: A lot of what that
happens in an organism is
based on chemical reactions.
SAMPLE ANSWER: Enzymes are
proteins that speed up

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like quality control
dosimetry, therapy planning
and other tasks that are
essential, but your
occupation environment might
be different, based on the
place you work. Your
capacity to earn is based on

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the amount of work you would
like to put in.

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Answers — Semesprit~~

The four main types of
carbon-based molecules in
organisms are carbohydrates,

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lipids, nucleic acids, and
starches. fatty acids.
monosaccharides. proteins.
Q. Both animal fats and
plant oils are made up of
glycerol and. phospholipids.
fatty acids.

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The Chemistry of Life ANSWER
KEY Section 2-1 The Nature
of Matter (pages 35-39) This
section identifies the three
particles that make up

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atoms. It also explains how atoms of the same element can have a different number of neutrons and describes the two main types of chemical bonds. Atoms (page 35) 1.

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Question: The Chemistry Of
Life EXERCISE 1: TESTING FOR
BIOMOLECULES Data Sheet
Table 1. Testing For
Proteins Results Sample
Initial Color Hypothesis
(Protein +/-) Final Color
Protein Present? Albumin (1)

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Gelatin (2) Glucose (3)

Water (4) E Table 2.

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of Matter (pages 35–39) This section identifies the three particles that make up atoms. It also explains how atoms of the same element can have a different number of neutrons and describes the two main types of

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Key Chapter 24 Atoms (page
35) 1.

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