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 $T = \frac{U}{S}$  (7)  $U = kT \ln \left( \frac{q+N}{q} \right)$  (8)  $S = k \left[ \ln(q+N) + 1 - \ln q \right]$  (9)  $U = kT \ln \frac{q+N}{q}$  (10)  $T = \frac{U}{k \ln \frac{q+N}{q}}$  (11)  $T = \frac{U}{k \ln \frac{q+N}{q}}$  (12) We can invert this to obtain  $U(T)$ :  $e^{-\frac{U}{kT}} = \frac{U}{U+N}$  (13)  $U = N e^{-\frac{U}{kT}}$  (14) Finally, the heat capacity is obtained from  $C = \frac{dU}{dT}$  (15)  $C = N e^{-\frac{U}{kT}} \frac{dU}{dT} = 2N e^{-\frac{U}{kT}} \frac{U}{kT}$  (16)  $C = 2N e^{-\frac{U}{kT}} \frac{U}{kT}$  (17)

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Step 1 of 7. (a) Use the values of Planck's constant and the speed of the light to find the value of  $h\nu$ . The Planck's constant is  $h$  and the speed of light is  $c$ . Thus, the value of  $h\nu$  is, Round off to three significant figures, the value of  $h\nu$  is . Therefore, the given result is proved. Comment ( 0) Step 2 of 7.

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